

Total synthesis of new steroids having an aromatic A ring with a 3-OH

Philippe Maurin, Malika Ibrahim-Ouali and Maurice Santelli*

Laboratoire de Synthèse Organique, UMR no 6009, Centre de St-Jérôme, Av. Esc. Normandie-Niemen, 13397 Marseille Cedex 20, France

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Abstract—A new strategy for introducing an hydroxy group at the position 3 of an aromatic A ring in a steroid, which is a key position for biological purposes, is described. This procedure has required a judicious choice of a *t*-butyl ether as protective group at the beginning of the synthesis, and its deprotection can be achieved in high yield in the final step. © 2001 Elsevier Science Ltd. All rights reserved.

Estrogens are active hormonal steroids possessing an aromatic A ring. It has been well established that estrogens act as important endocrine growth factors for at least a third of breast cancers.¹

So we were interested in developing new molecules which might be used for the treatment of estrogen-dependent diseases. The A ring of those steroids is in general a phenol with an hydroxy group at the 3-position, required for the binding to the estrogen receptor and for inhibiting the 17- β hydroxysteroid dehydrogenase.²

In previous papers,³ we reported a novel strategy for the preparation of a wide variety of steroids, which involved the addition of 1,8-bis(trimethylsilyl)-2,6-octadiene (BISTRO) to various electrophilic reagents.⁴

All the steroids previously described in our synthetic work possess an aromatic A ring substituted by a methoxy group at the 2- and/or 3-position. Nevertheless, all attempts realized for the final demethylation to obtain the required hydroxy group failed using different conditions (BBr₃ in CH₂Cl₂,⁵ DIBALH in toluene,⁶ CH₃SO₃H).

So we now settle this problem by using a t-butyl ether as a protective group, easier to remove than the precedent methyl ether.⁷

Keywords: intramolecular Diels-Alder; Wacker-type oxidation; protective group.

The required precursor steroids were prepared in the same way we used in our previous work (Scheme 1).³

First, using a procedure analogous to that described by Kametani⁸ and Honda,⁹ we efficiently prepared the 1-iodo-4-tert-butoxybenzocyclobutene 1. The condensation of the activated spirolactone 2^{10} with 1, using K₂CO₃ in acetone, led to the benzocyclobutene 3 in 72% yield. Compound 3 was then submitted to a decarbomethoxylation according to the Krapcho procedure (NaCN/DMSO). 11 Heating the new benzocyclobutene intermediates 4 afforded a mixture of two cycloadducts 5a and 5b in 68% yield and a 2:1 ratio. The two steroids 5a and 5b were separable by chromatography over silica gel and have, respectively, a trans-anti-cis and a trans-anti-trans ring fusion. 12 Much to our surprise and in contrast with the results previously observed for the steroids possessing a 3-methoxy group, the main product 5a matches here the trans-anti-cis ring fusion of non-natural products.

The Wacker-type oxidation¹³ of the vinyl group of **5b** gave the corresponding ketone **6** and the aldehyde **7** resulting from an anti-Markovnikov hydroxypalladation in a 10:1 ratio and a 56% overall yield (Scheme 2). The formation of the minor product has been rationalized by an intramolecular coordination of the palladium with the oxygen of the lactone moiety.¹⁴

The final step of our work, which constituted a key step, was the deprotection of the 3-OH. Preliminary experiments were carried out on compound **5b** and removal of the *tert*-butyl protective group could be achieved under mild conditions TFA:CH₂Cl₂ 1:1,^{7b,15} to afford **8** in 66% yield (Scheme 3).

^{*}Corresponding author. Fax: (33) 4 91 98 38 65; e-mail: m.santelli@lso.u-3mrs.fr

Scheme 1.

Scheme 2.

Scheme 3.

The same conditions applied to **6**, which possess the stereochemistry of the natural products, gave **9** in 83% yield (Scheme 4). This molecule can be considered as a good intermediate for preparing new compounds and examining their estrogenic properties.

Conclusion: We have shown herein that our strategy is compatible with the synthesis of a wide range of steroids and particularly steroids possessing an hydroxy group in a biologically important position in the aromatic A ring. We are currently pursuing our research towards more complex target systems.

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Scheme 4.

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